

# Radioactivity

Lecture 23

Radioactivity and Art Analysis

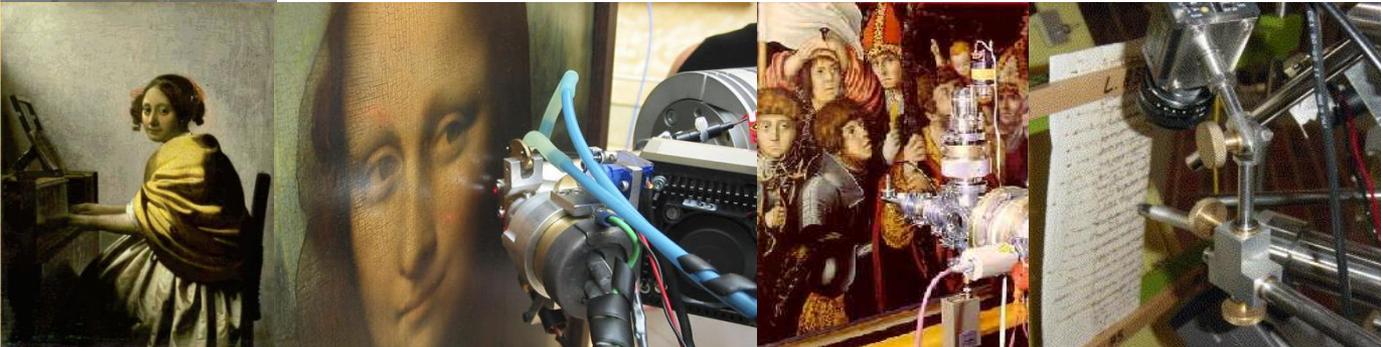
# Modern Tools for Ancient Art



Modern art analysis techniques rely on the quantum nature of matter to determine provenance, age, techniques, and forgeries.



The most frequently used methods are x-ray analysis such as PIXE and XRF, coupled with atomic analysis techniques such as Raman spectroscopy, and nuclear physics techniques such as Neutron activation analysis.



# The origin of materials

XRF analysis with portable instrumentation

The “Relics of the three Magi”, came from Milan, Italy to Cologne, Germany in 1162. The shrine was made in 1180-1225 AD. Where did jewels, gems, and other precious materials come from?

# Analysis of paint pigments

Pre 1800 oil paintings contained specific pigments prepared from naturally available materials to achieve color effects. After 1850 these pigments were gradually replaced by organic (Carbon based) pigments provided by the chemical industry.

## White pigments

Antimony white	$\text{Sb}_2\text{O}_3$
Lithopone	$\text{ZnO} + \text{BaSO}_4$
Permanent white	$\text{BaSO}_4$
Titanium white	$\text{TiO}_2$
White lead	$2\text{PbCO}_3 \cdot \text{Pb(OH)}_2$
Zinc white	$\text{ZnO}$
Zirconium oxide	$\text{ZrO}_2$
Chalk	$\text{CaCO}_3$
Gypsum	$\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$

## Yellow pigments

Auripigmentum	$\text{As}_2\text{S}_3$
Cadmium yellow	$\text{CdS}$
Chrome yellow	$2\text{PbSO}_4 \cdot \text{PbCrO}_4$
Cobalt yellow	$\text{K}_3[\text{Co(NO}_2)_6] \cdot 1.5\text{H}_2\text{O}$
Lead-tin yellow	$\text{Pb}_2\text{SnO}_4 / \text{PbSn}_2\text{SiO}_7$
Massicot	$\text{PbO}$
Naples yellow	$\text{Pb(SbO}_3)_2 / \text{Pb}_3(\text{SbO}_4)_2$
Strontium yellow	$\text{SrCrO}_4$
Titanium yellow	$\text{NiO} \cdot \text{Sb}_2\text{O}_3 \cdot 20\text{TiO}_2$
Yellow ochre	$\text{Fe}_2\text{O}_3 \cdot n\text{H}_2\text{O}$ (20–70%)
Zinc yellow	$\text{K}_2\text{O} \cdot 4\text{ZnO} \cdot 4\text{CrO}_3 \cdot 3\text{H}_2\text{O}$

## Red pigments

Cadmium red	$\text{CdS} + \text{CdSe}$
Cadmium vermilion	$\text{CdS} + \text{HgS}$
Chrome red	$\text{PbO} \cdot \text{PbCrO}_4$
Molybdate red	$7\text{PbCrO}_4 \cdot 2\text{PbSO}_4 \cdot \text{PbMoO}_4$
Realgar	$\text{As}_2\text{S}_3$
Red lead	$\text{Pb}_3\text{O}_4$
Red ochre	$\text{Fe}_2\text{O}_3$ (up to 90%)
Vermilion	$\text{HgS}$

## Green pigments

Basic copper sulfate	$\text{Cu}_x(\text{SO}_4)_y(\text{OH})_z$
Chromium oxide	$\text{Cr}_2\text{O}_3$
Chrysocolla	$\text{CuSiO}_3 \cdot n\text{H}_2\text{O}$
Cobalt green	$\text{CoO} \cdot 5\text{ZnO}$
Emerald green	$\text{Cu}(\text{CH}_3\text{COO})_2 \cdot 3\text{Cu}(\text{AsO}_2)_2$
Guignent green	$\text{Cr}_2\text{O}_3 \cdot n\text{H}_2\text{O} + \text{H}_3\text{BO}_3$
Malachite	$\text{CuCO}_3 \cdot \text{Cu(OH)}_2$
Verdigris	$\text{Cu}(\text{CH}_3\text{COO})_2 \cdot n\text{Cu(OH)}_2$

## Blue pigments

Azurite	$2\text{CuCO}_3 \cdot \text{Cu(OH)}_2$
Cerulean blue	$\text{CoO} \cdot n\text{SnO}_2$
Cobalt blue	$\text{CoO} \cdot \text{Al}_2\text{O}_3$
Cobalt violet	$\text{Co}_3(\text{PO}_4)_2$
Egyptian blue	$\text{CaO} \cdot \text{CuO} \cdot 4\text{SiO}_2$
Manganese blue	$\text{BaSO}_4 \cdot \text{Ba}_3(\text{MnO}_4)_2$
Prussian blue	$\text{Fe}_4[\text{Fe(CN)}_6]_3$
Smalt	$\text{Co-glass} (\text{K}_2\text{O} + \text{SiO}_2 + \text{CoO})$
Ultramarine	$\text{Na}_{8-10}\text{Al}_6\text{Si}_6\text{O}_{24}\text{S}_{2-4}$

## Black pigments

Antimony black	$\text{Sb}_2\text{O}_3$
Black iron oxide	$\text{FeO} \cdot \text{Fe}_2\text{O}_3$
Carbon or charcoal black	$\text{C}$ (95%)
Cobalt black	$\text{CoO}$
Ivory black	$\text{C} + \text{Ca}_3(\text{PO}_4)_2$
Manganese oxide	$\text{MnO} + \text{Mn}_2\text{O}_3$

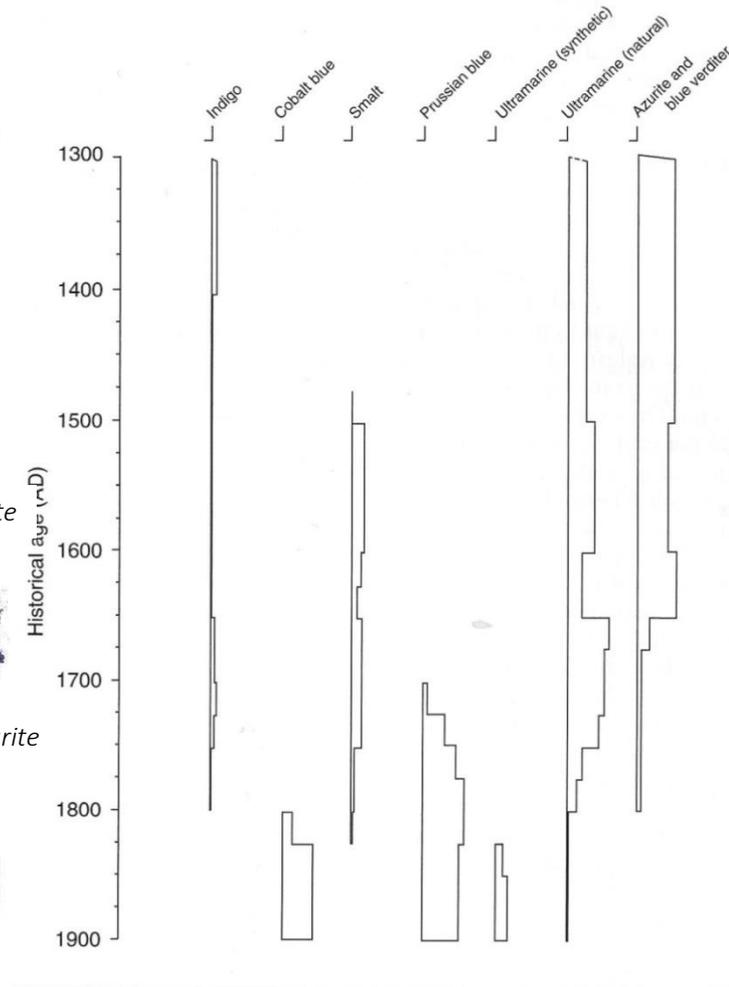
# Pigments available until 1800 AD

Paint is composed of a colored pigment and a binder substance

Pigment: colored powdered substance grinded from minerals salts, or dyes

Binder: Material that evenly disperses the pigment, adheres to surface when paint applied and then dries.

Paints are throughout uniform homogeneous mixtures.



# X-Ray Fluorescence of Manuscripts

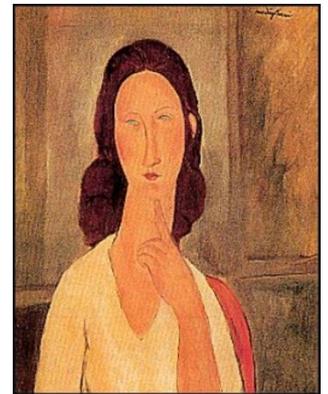
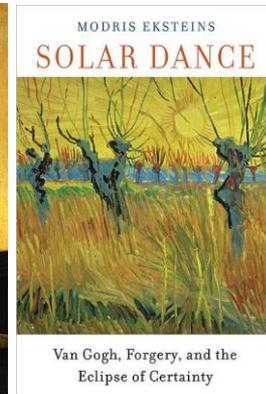


# Opportunities in Art Forgery

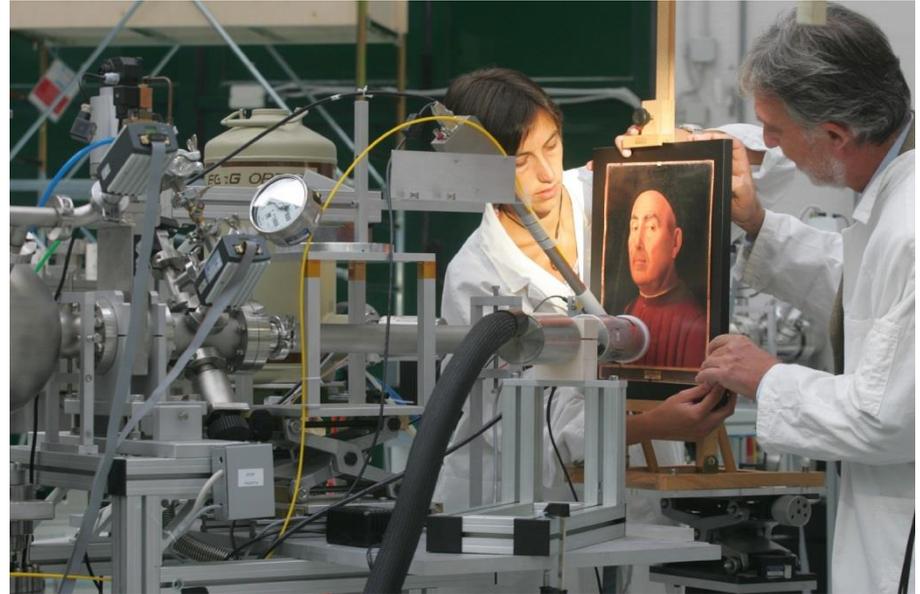
Science techniques are an emerging tool for:

- Forgery analysis by nuclear forensic techniques in a competitive art market (Vermeer, Van Gogh, Modigliani, Rothko, etc)

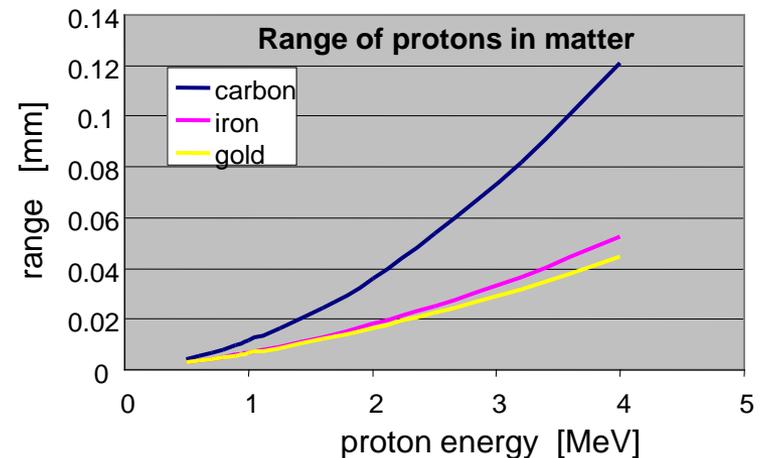
e.g. Vermeer forgeries by Hans van Meegeren  
Van Gogh forgeries by Otto Wacker  
>1000 Modigliani fakes by Elmyr de Hory  
New York galleries sold Mark Rothko,  
Jackson Pollock and Willem de Kooning  
forgeries; damage unknown



# Proton Induced X-ray Emission (PIXE)

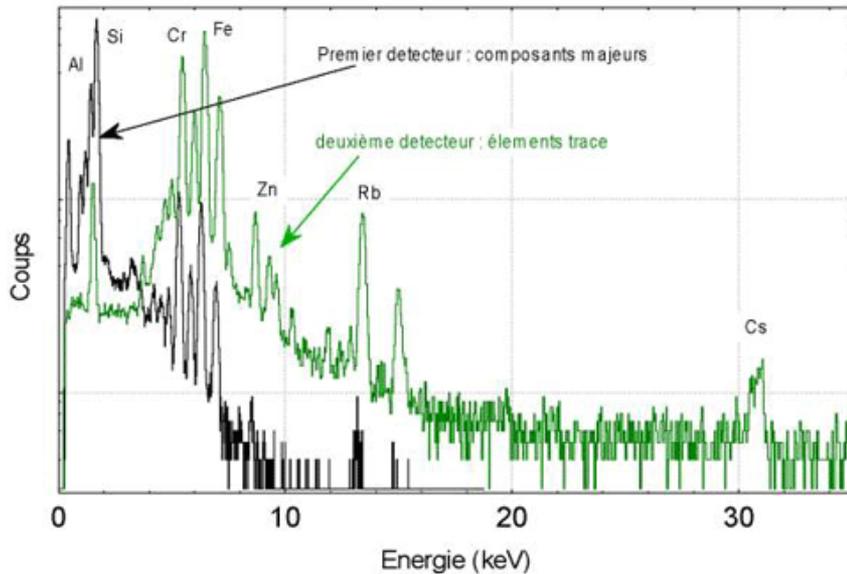


Accelerated particles like protons penetrate deeper into material, which reduces distorting surface effects that handicap XRF analysis. Depending on energy the composition of deeper material layers can be explored. That provides an additional insight in chemical decomposition processes important for restoration procedures.



# Tracing Material Origins

The red stone eyes of the statue of the Parthian goddess of love Ishtar were originally thought by Louvre curators to be made of colored glass



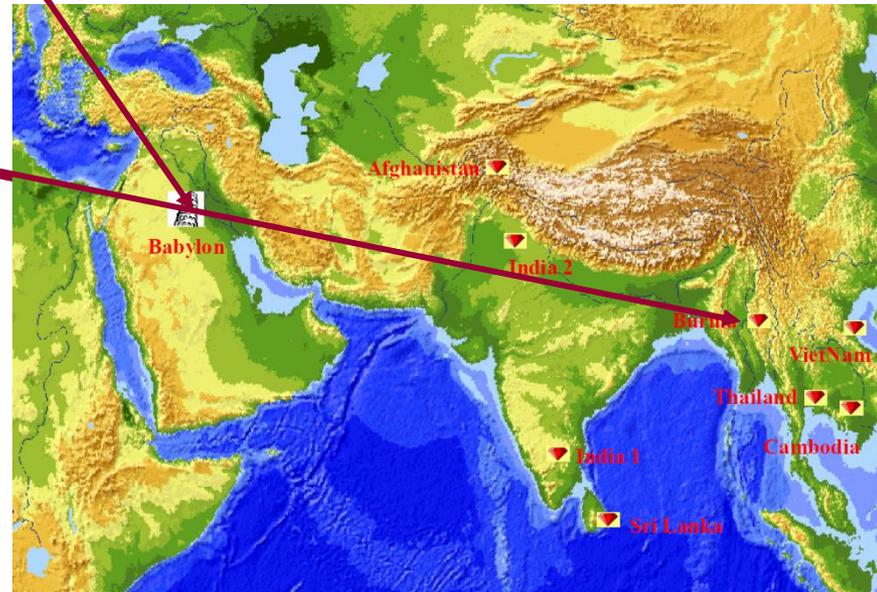
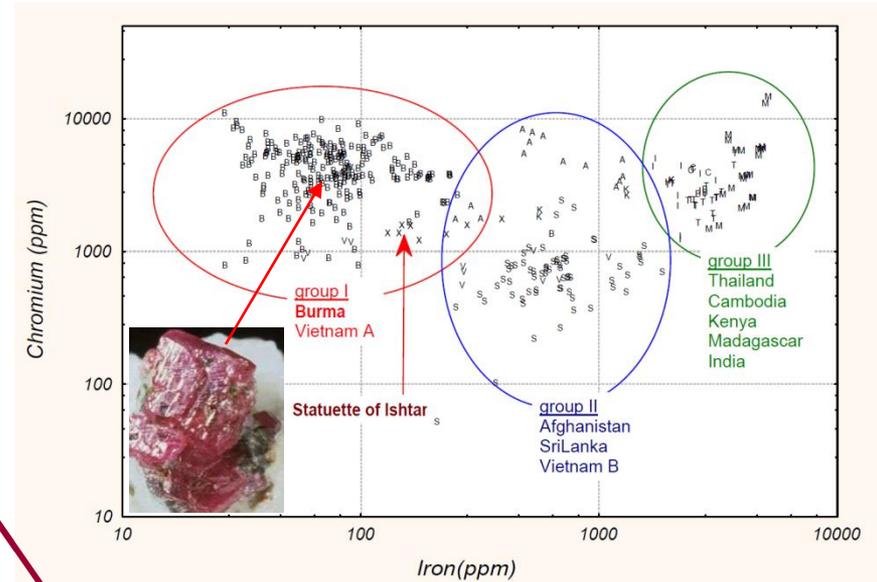
PIXE analysis showed that the inlays were rubies.  $Al_2SiO_4(F,OH)_2 + (Cr,Fe \text{ rich})$



# Provenance, or where did the rubies come from?

The trace element content provides the fingerprint of provenance in archaeology

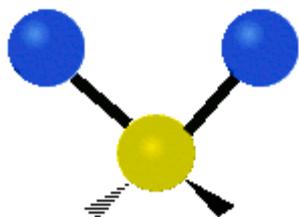
Comparison of Fe versus Cr content in the Ishtar rubies found in Mesopotamia with rubies from various provenances shows strong indication that rubies did originate from Burma. Ancient trade connections (silk road) between near and far east empires!



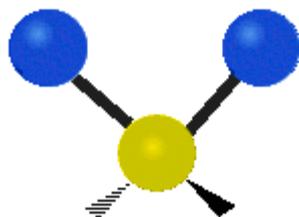


# Raman Spectroscopy of Molecules

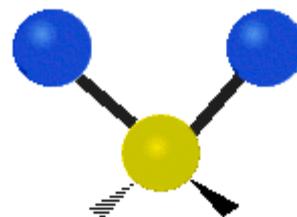
Provides a spectroscopic tool for analyzing molecular components in pigments by looking for signals corresponding to molecular excitation modes (vibration, rotation and combinations of such). Raman Spectroscopy is therefore also tool for analysis of modern organic chemistry based pigments.



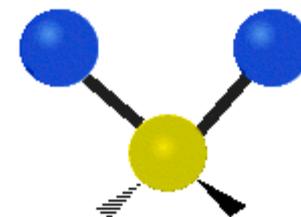
stretching



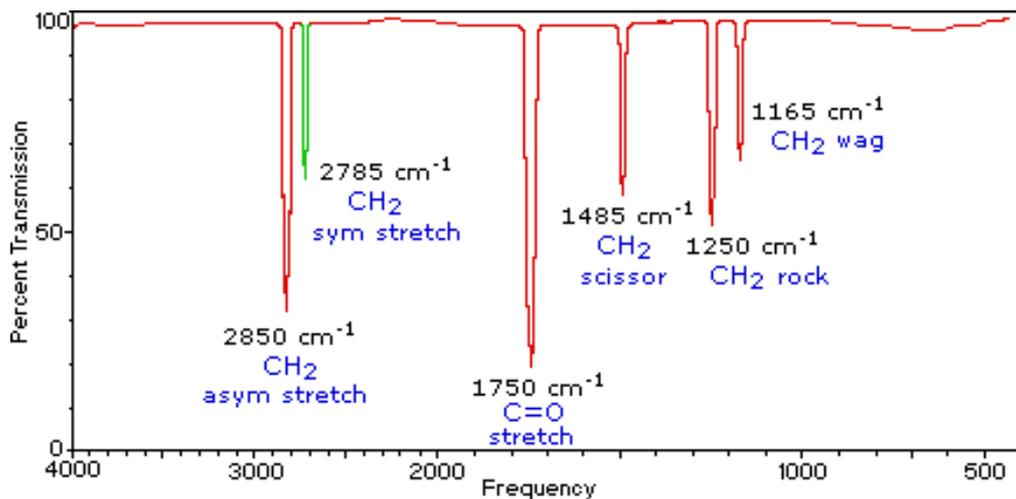
bending



scissoring



twisting



Excitation source for excitation process is a monochromatic laser. Raman Spectroscopy works in the infrared since molecular excitations are less energetic than atomic or nuclear excitations.

# Testing ink pigments of medieval monastery handwriting of letter **R**

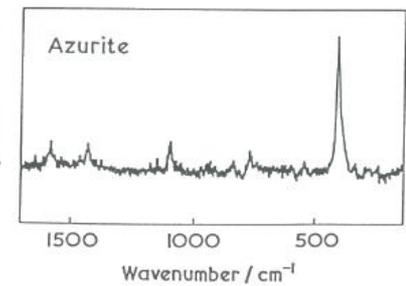
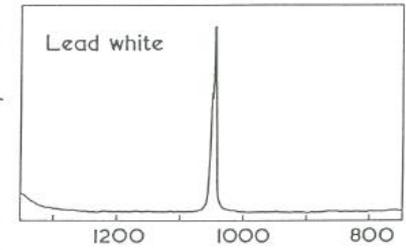
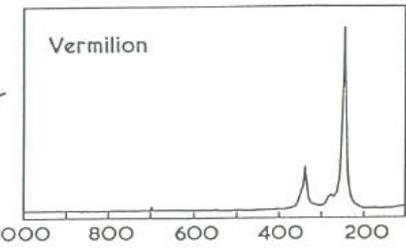
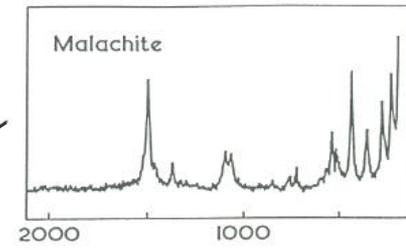
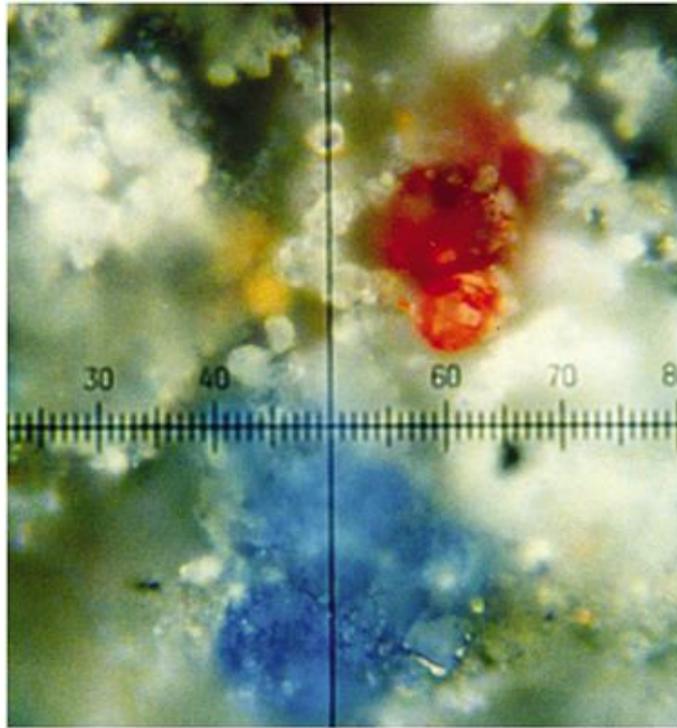
Lead white:  $k=1050\text{ cm}^{-1}$  ( $\text{PbCO}_3$ )

Malachite:  $(\text{Cu}^{2+}_2(\text{CO}_3)(\text{OH})_2)$

Azurite:  $(\text{Cu}^{2+}_3(\text{CO}_3)_2(\text{OH})_2)$

Vermillion:  $k=253\text{ cm}^{-1}$   $285\text{ cm}^{-1}$ ,  $343\text{ cm}^{-1}$  ( $\text{HgS}$ ) (cinnabar)

Minium:  $k=226\text{ cm}^{-1}$ ,  $313\text{ cm}^{-1}$ ,  $390\text{ cm}^{-1}$ ,  $549\text{ cm}^{-1}$  ( $\text{Pb}_2\text{O}_3$ )

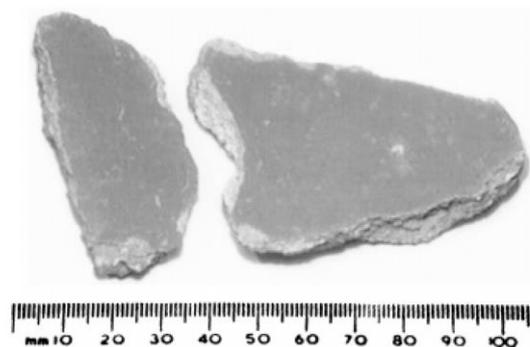


# Frescoes in Herod's Tomb in Jericho

Roman fresco technique: lime wash, followed by pigment application

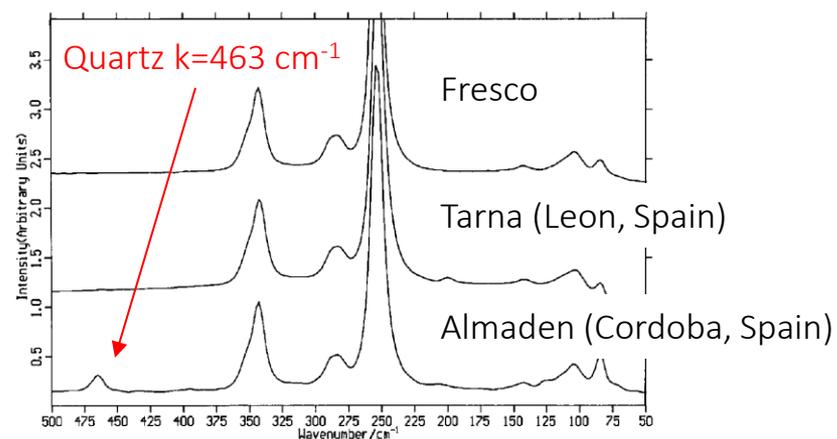
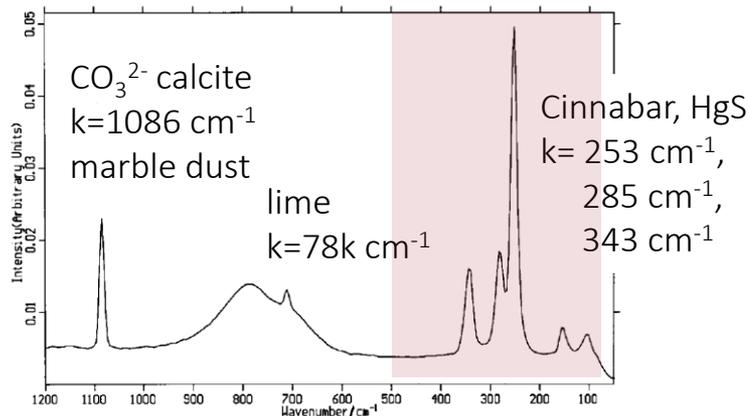


Analysis of fragments with Raman spectroscopy



Cinnabar (Persian Dragon's blood):  
HgS (vermilion)

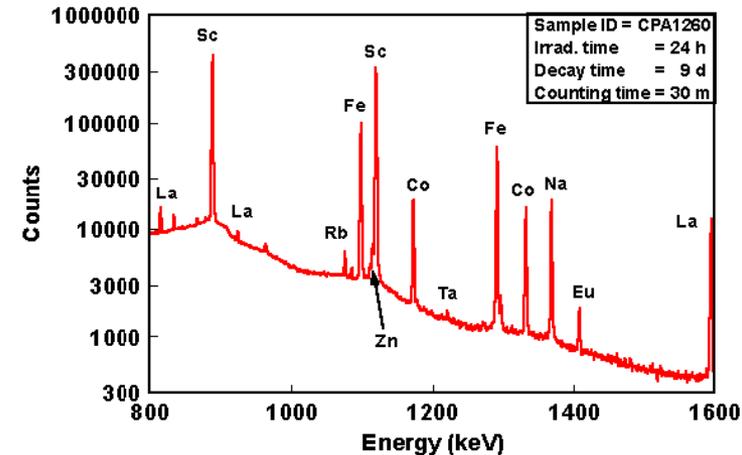
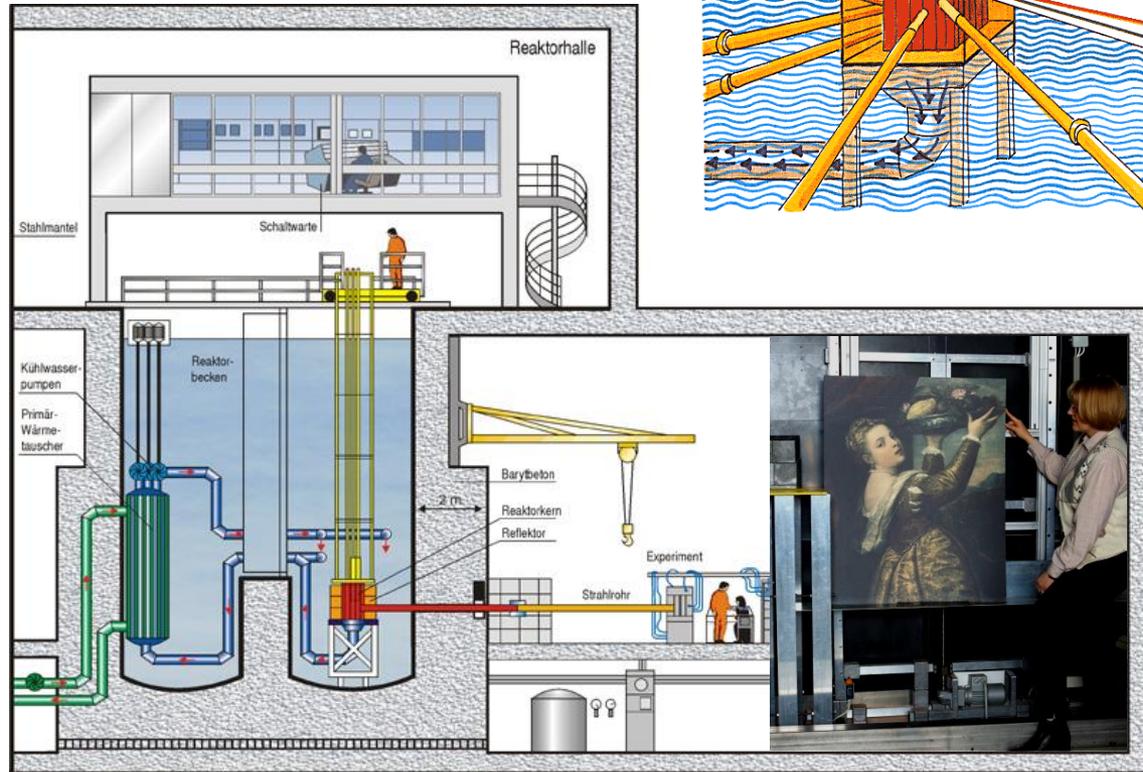
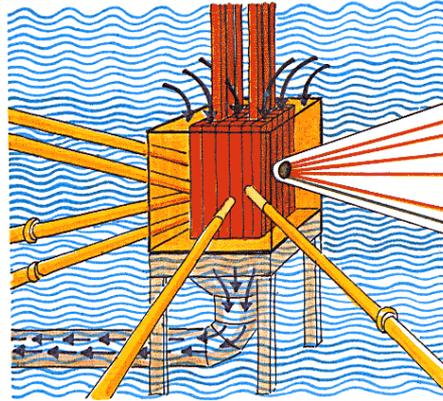
Provenance of HgS pigment  
(Pliny & Vitruvius claim Spain)



1064 nm excitation

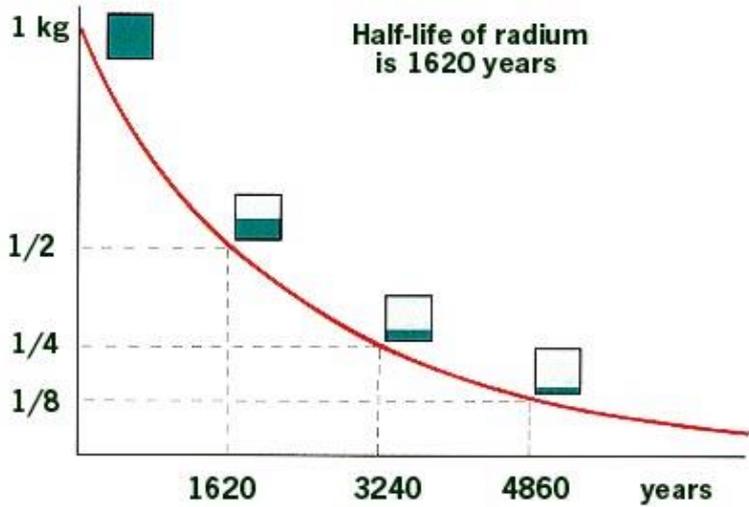
# Neutron Activation (NA)

Expose material to high neutron flux and add neutrons to nuclei to produce an radioactive isotope with subsequent analysis of its characteristic radioactive decay pattern.



# Timescale and Radiation Sensitivity

Signatures are either characteristic radiation or characteristic decay time, which is different for each radioactive isotope



Chemical element	Associated pigment	Radioactive isotope formed during activation and its half-life	Time period after activation during which best images in autoradiographs are produced
manganese	umber, dark ocher	Mn <sup>56</sup> , 2.6 hours	0–24 hours
copper	malachite, azurite, verdigris	Cu <sup>66</sup> , 5.1 minutes Cu <sup>64</sup> , 12.8 hours	0–20 minutes 1–3 days
sodium	glue, medium, canvas, ultramarine	Na <sup>24</sup> , 15.0 hours	1–3 days
arsenic	smalt, glass	As <sup>76</sup> , 26.5 hours	2–8 days
phosphorus	bone black	P <sup>32</sup> , 14.3 days	8–30 days
mercury	vermilion	Hg <sup>203</sup> , 48 days	more than 25 days
cobalt	smalt, glass	Co <sup>60</sup> , 5.3 years	more than 25 days

Taking advantage of radioactive decay

Table 1. Chemical elements and associated pigments most frequently observed in autoradiography of seventeenth-century Dutch and Flemish paintings.

The following pigments generally do not cause distinct images in autoradiographs: chalk, lead white, ocher, lead-tin yellow, lakes, madders, and indigo.

# The Man with the Gold Helmet

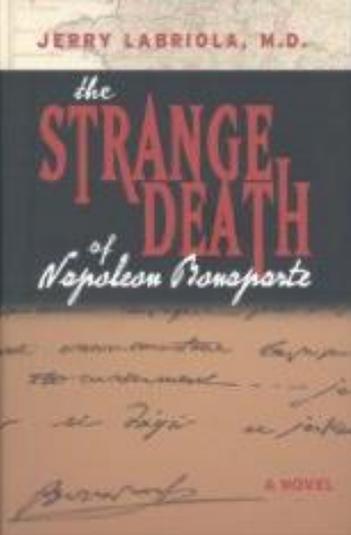
by Rembrandt van Rijn?



# Was Napoleon murdered by the British?

Neutron activation comes handy

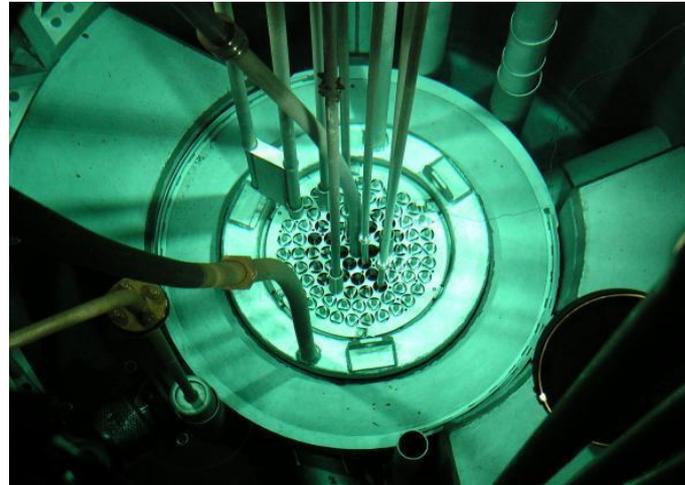
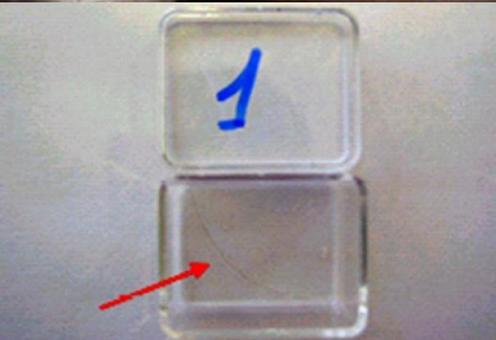




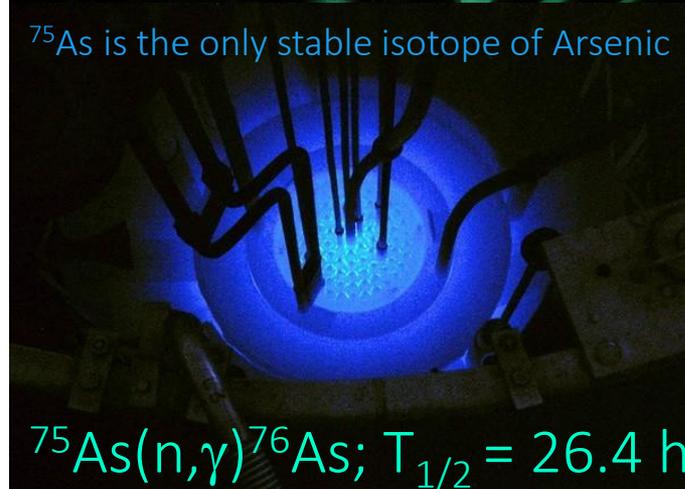
# Napoleon's Death

May 5 1821

poisoned by Arsenic?????



$^{75}\text{As}$  is the only stable isotope of Arsenic



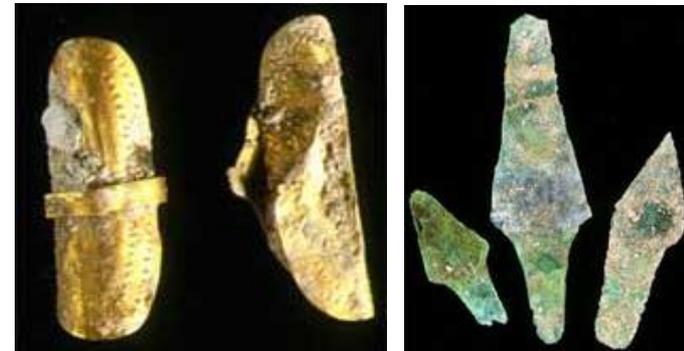
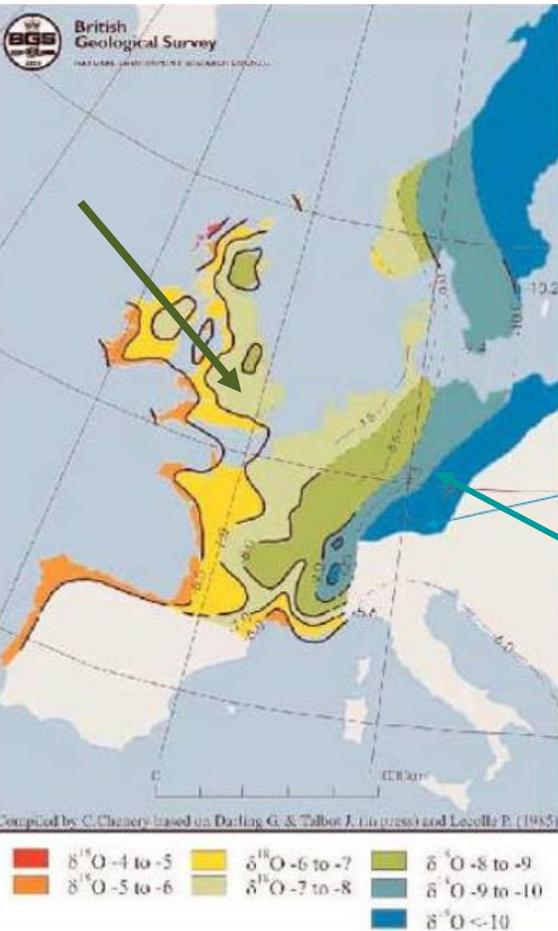
$^{75}\text{As}(n,\gamma)^{76}\text{As}; T_{1/2} = 26.4 \text{ h}$

Napoleon has declared in his will that 'I die before my time, murdered by the English oligarchy and its hired assassin'.

The Emperor's hair had an average arsenic level of around 10-15 ppm, whereas the arsenic level in the hair samples from currently living persons is around 0.1 ppm. **But surviving relatives had similar levels!**

# Stable Isotope Analysis (SIA), the King of Stonehenge at 2300 BC

Chemophysical fractionation of isotopes cause local changes in abundance ratio. Climate and rain pattern influence the  $^{18}\text{O}$  to  $^{16}\text{O}$  isotope ratio from sea to land.



The Daily Express expressed the opinion  
*"This is as shocking as the discovery that the first cricket players wore leather pants and ate Bratwurst with their tea".*

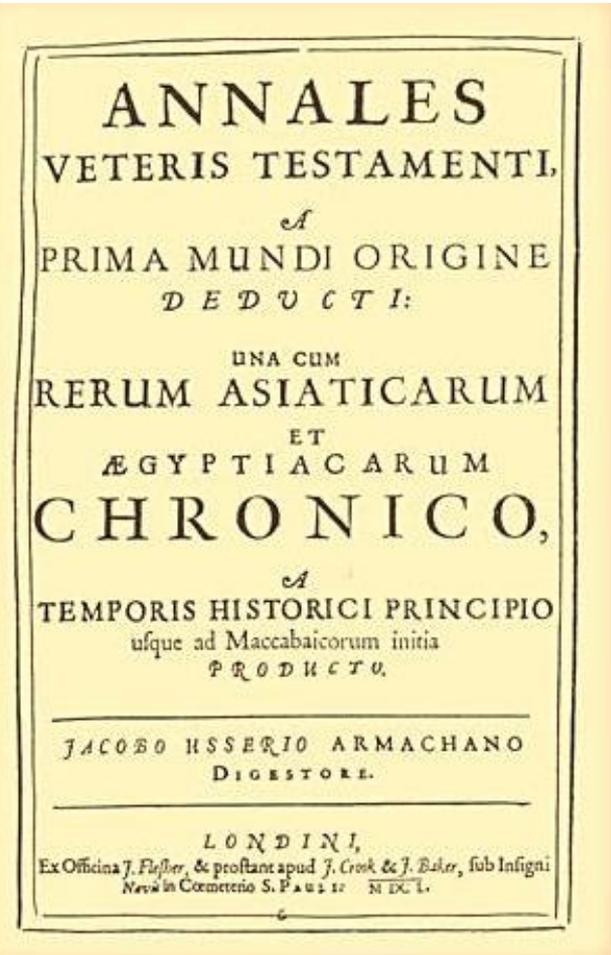
# Archaeological Dating the past

*“Everything which has come down to us from heathendom is wrapped in a thick fog; it belongs to a space of time we cannot measure. We know that it is older than Christendom, but whether by a couple of years or a couple of centuries, or even by more than a millennium, we can do no more than guess”*

*Rasmus Nyerup, 1802*

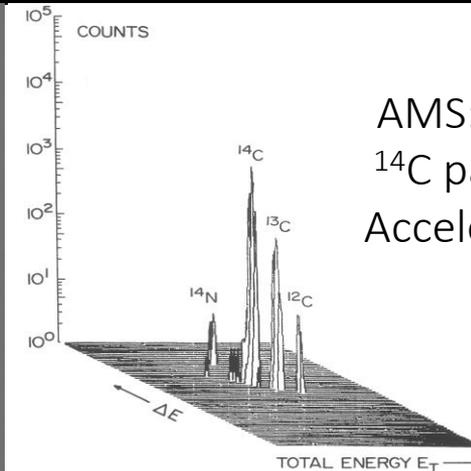
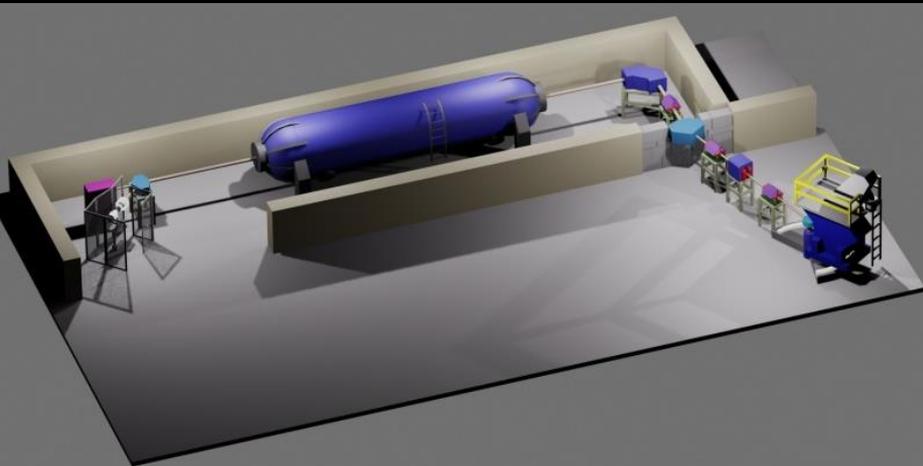
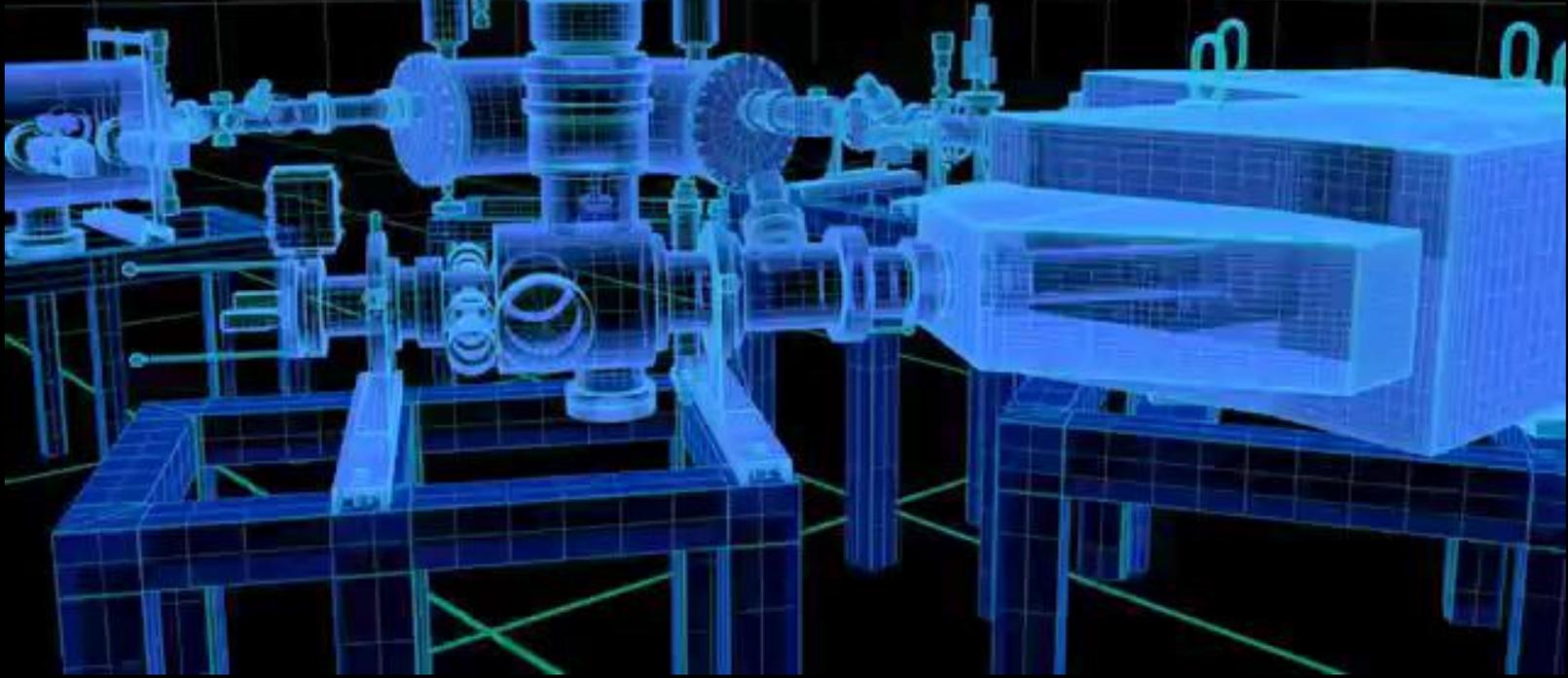


# Archaeological clocks



How to measure the time and age of things?

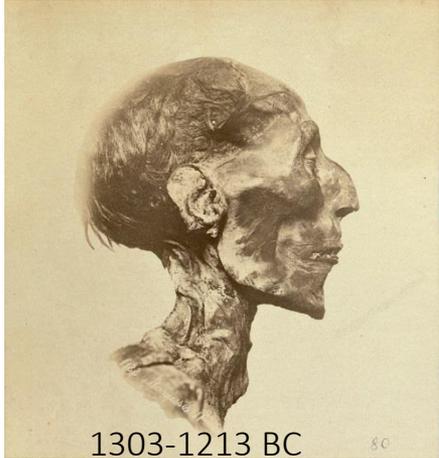
# $^{14}\text{C}$ dating with AMS



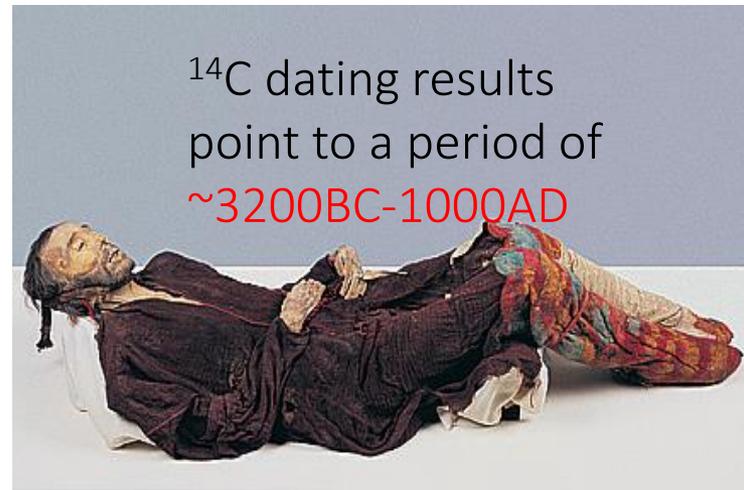
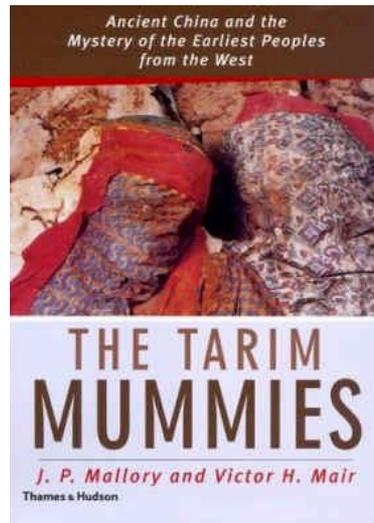
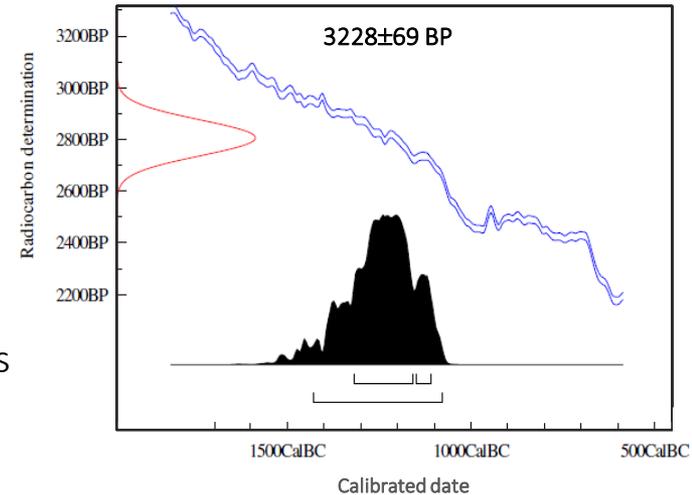
AMS: counting the radioactive  $^{14}\text{C}$  particles with accelerators:  
Accelerator Mass Spectrometry

# Dating Mummies

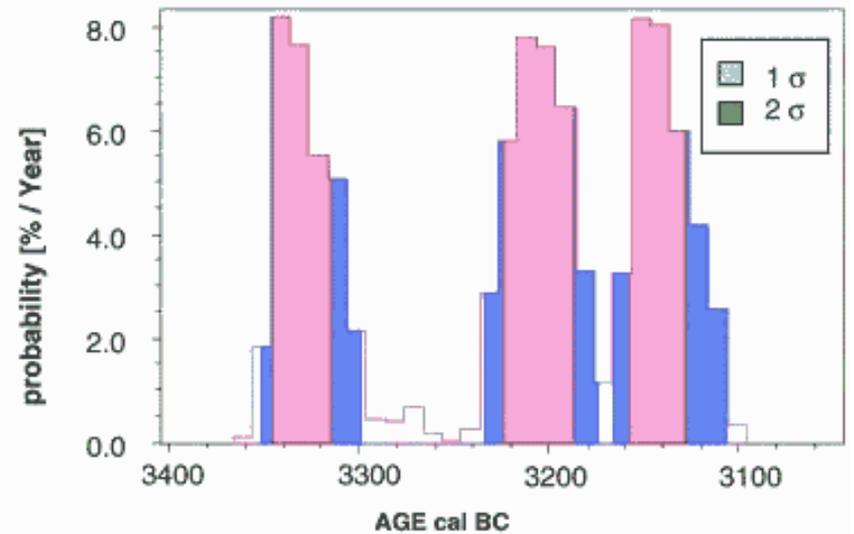
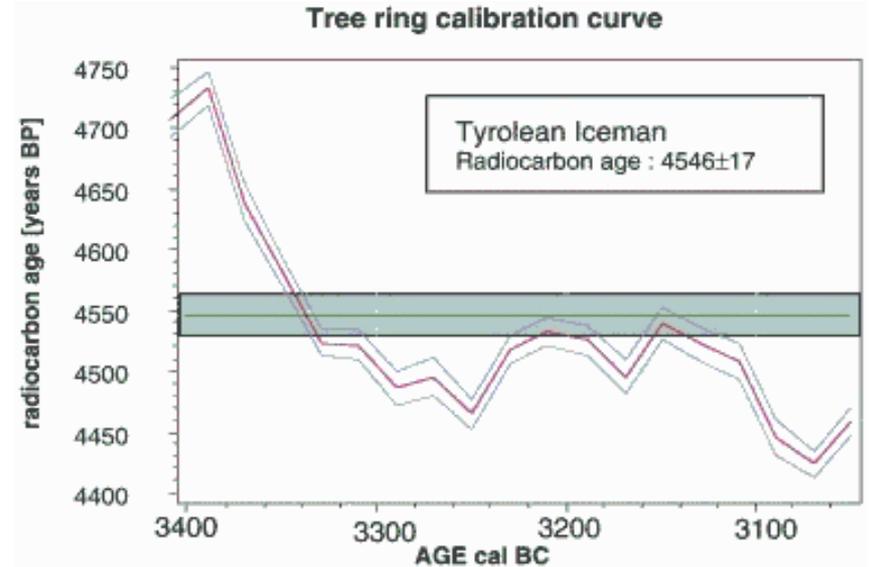
'My name is Ozymandias, king of kings:  
Look on my works, ye Mighty, and despair!'



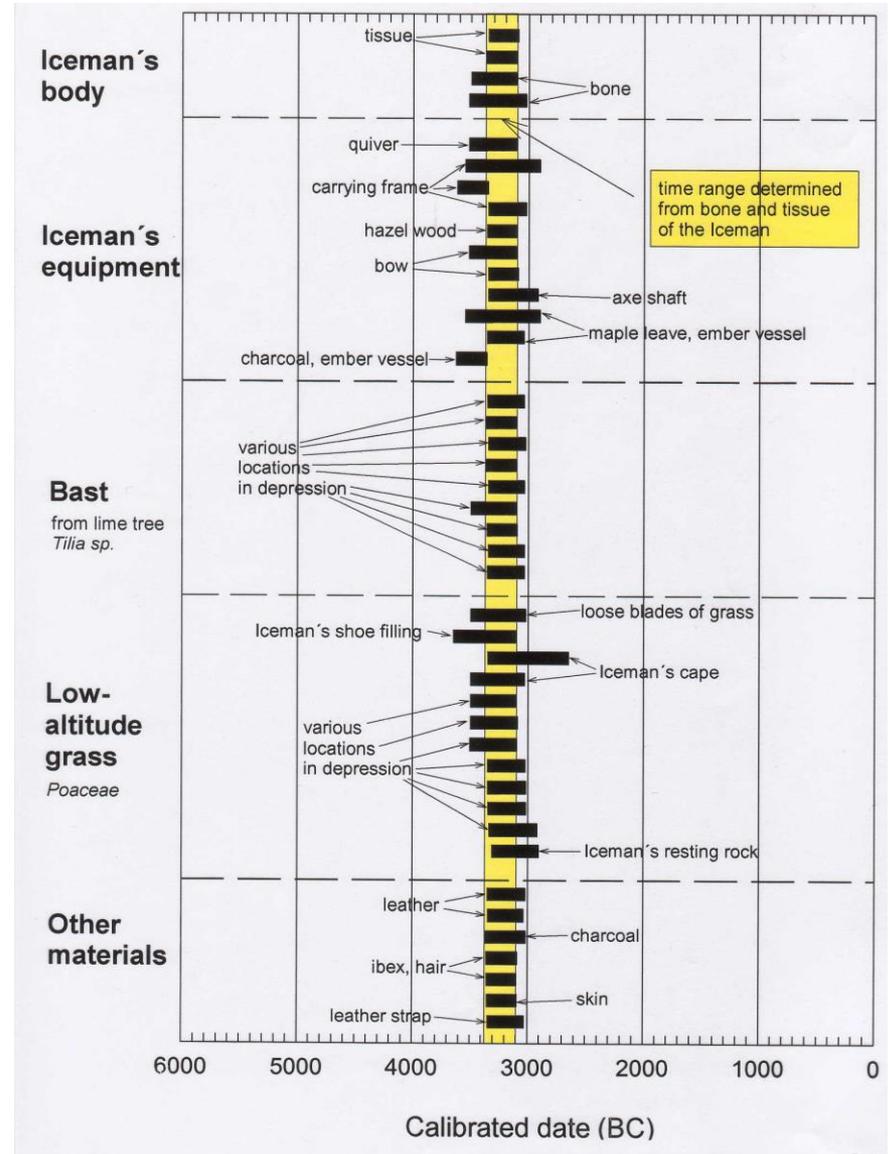
The mummy of Ramses II was one of the first samples tested by the new  $^{14}\text{C}$  radiocarbon method to check the reliability of Egyptian dynasty counting versus biblical counting.



# Conserved by ice - Oetzi, the iceman



# Murder 5000 years ago

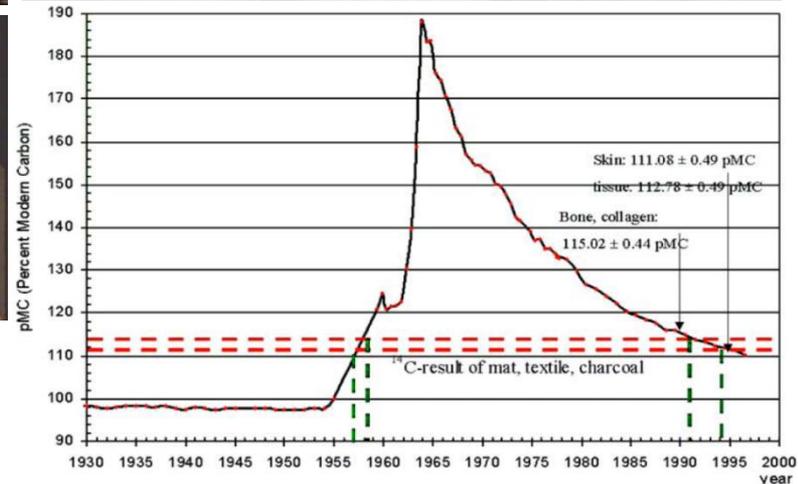


# The big business with (false) artifacts

Police raid of an art dealer in Karachi, Pakistan in October 2000 found a mummy, supposedly the daughter of Cyrus the Great (576-530 BC). The asking price of the dealer was US \$ 11 Million.



Owner claims were filed by the governments of Pakistan, Iran, Afghanistan (Taliban).



AMS analysis determined a large <sup>14</sup>C amount in the mummy and dated her death to ~1993 !

# Tracking Illegal Ivory Trade



❖ Increasing slaughter of elephants since 1970 with increased use of automatic weapons.

❖ Ivory trade ban in 1989 to protect elephants from becoming extinct

❖ Growth in poaching and smuggle leading to local decline of elephant population as high as 90%

# Lab-Tour Overview

TwinSol radioactive  
beam production

Gas-filled spectrometer  
for AMS applications

5 MV single ended Pelletron for  
light and heavy ion beams  
ECR source in terminal

11MV Pelletron Tandem  
Accelerator for nuclear  
structure measurements,  
AMS applications, and  
PIXE analysis.

St. George recoil separator  
for low energy inverse kinematics  
experiments

